

A NOVEL PREGNANE GLYCOSIDE FROM *PERIPLOCA CALOPHYLLA*

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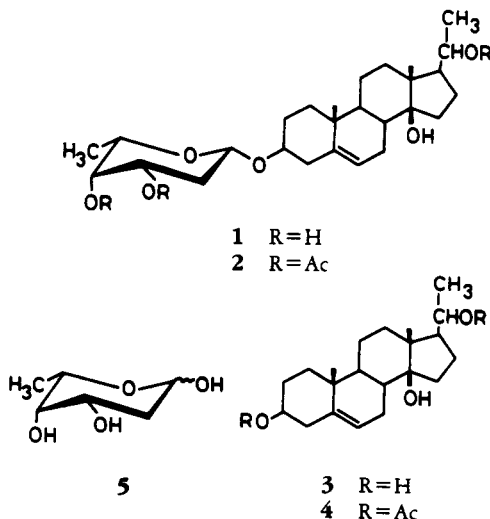
Pregnane derivatives and cardenolides have been reported to be present in several species of the Asclepiadaceae family (1-4). In earlier investigations of the twigs of *Periploca calophylla* Falc. four novel glycosides, locin (5), plocin (6), plocinin (7), and calocin (8), as well as triterpenoids (9) were reported. We now report the isolation of the new glycoside calocinin [1].

Calocinin [1], $C_{27}H_{44}O_6$, mp 250-255°, $[\alpha]_D + 16^\circ$, isolated from the combined $CHCl_3$ -EtOH (4:1 and 3:2) extracts of the twigs of *P. calophylla*, responded positively to the Liebermann-Burchardt test (10), to xanthhydrol (11, 12), and to the Keller-Kiliani reaction (13), indicating it to be a steroidal glycoside of a 2-deoxy sugar. In addition, characteristic methylene signals in the region δ 2.25-2.14 (1H) and 1.72-1.68 (1H) along with a secondary methyl doublet at δ 1.34 ($J = 6$ Hz) and an anomeric proton at δ 4.64 (dd, $J = 10$ and 2.5 Hz) in the 1H nmr spectrum of 1 suggested it to be a monoglycoside of 2,6-dideoxy hexose.

Mild acid hydrolysis of calocinin [1] with 0.025 M H_2SO_4 (14) afforded a crystalline genin 3, $C_{21}H_{34}O_3$, mp 198-201°, $[\alpha]_D - 50^\circ$, and the sugar 5 as a viscous syrup, $[\alpha]_D - 58.5^\circ$. A measurement of the relative mobility of 5 using digitoxose as a reference (15) indicated it to be 2,6-dideoxy-L-fucose (2,6-dideoxy-L-lyxo hexose) (16), and this assignment was confirmed by oxidation and conversion of the resulting lactone to its phenylhydrazide, identical with L-fuconic acid phenylhydrazide (17).

The pregnane genin 3 (C_{21}) was characterized as calogenin ($3\beta, 14\beta, 20\epsilon$ -trihydroxypregn-5-ene) by co-chromatography (tlc) with an authentic sample. This finding was confirmed by preparation of its diacetate 4, $C_{25}H_{38}O_5$, mp 163-165°, $[\alpha]_D - 18^\circ$, identical with di-O-acetylcalogenin (mmp, $[\alpha]_D$, tlc).

The eims of the glycoside 1 did not exhibit a molecular ion. The highest peak at m/z 419 corresponded to the loss of the C-17 hydroxyethyl chain, indicating that the sugar was glycosidically linked to the C-3 hydroxyl group and



not to the C-20 hydroxyl group of calogenin. The spectrum also showed peaks for the genin moiety at m/z 334, 289 [$334 - \text{CH}(\text{OH})\text{CH}_3$] $^+$, 271 [$289 - \text{H}_2\text{O}$] $^+$, 253 [$289 - 2\text{H}_2\text{O}$] $^+$, accounting for all three oxygen functions in the genin. The base peak at m/z 131 for the fucosyl cation was accompanied by peaks at m/z 113, 95, 72, and 68 which could be interpreted as arising from the characteristic fragmentation pathway of 2,6 dideoxy hexose (18).

The ^1H -nmr spectrum of the glycoside **1** at 400 MHz not only confirmed the derived structure but also established the configuration of the glycosidic linkage. The large coupling constant ($J = 10$ Hz) of the anomeric proton double doublet at δ 4.64 attributed to its axial configuration suggested that the sugar was present in $^1\text{C}_4$ (L) conformation and linked to the genin through a β -glycosidic linkage. The two singlets at δ 1.24 (3H) and δ 1.02 (3H), a one-proton quartet at δ 3.75 ($J = 6$ Hz), and two three-proton doublets centered at δ 1.30 ($J = 6$ Hz) and δ 1.34 ($J = 6$ Hz) were attributed to the two tertiary C-methyl groups (C-18 and C-19), carbinol methine proton at C-20, and to the C-21 secondary methyl group of C-17 hydroxyethyl chain in the pregnane genin and C-6' secondary methyl group in the 6-deoxy hexose moiety, respectively. A triplet at δ 3.15 (1H) and two multiplets (1H each) centered at δ 3.32 and δ 5.32 were assigned to protons at C-4' and C-5' of the sugar moiety and C-6 of the genin moiety, respectively.

Nmnr experiments were helpful in the confirmation of these assignments. Irradiation of the double doublet at δ 4.64 corresponding to the C-1 anomeric proton caused the collapse of the multiplicity of the methylene signal in the region δ 2.25–2.14 and 1.72–1.68. Similarly, irradiation of the two doublets at δ 1.30 and δ 1.34 corresponding to the C-21 and C-6' methyl group caused collapse of the C-20 methine proton quartet into a singlet at δ 3.75 and the C-5'

methine proton multiplet into a doublet at δ 3.32, respectively. Conversely, irradiation of the C-20 carbinol methine proton quartet at δ 3.75 resulted in the collapse of doublet into a singlet at δ 1.30. These results indicated that the sugar was a 2,6-dideoxy hexose, and the pregnane genin bears a C-17 hydroxyethyl chain.

Acetylation of **1** with Ac_2O in pyridine yielded an amorphous triacetate **2**, $[\alpha]_{\text{D}}^{+12}$, also characterized from its 400-MHz ^1H nmr spectrum which consisted of signals for three acetyl groups as 3H singlets at δ 2.0, 2.02, and 2.05 in addition to two tertiary methyl group singlets at δ 0.9 and 1.0 ppm. Besides this, the C-20 methine proton quartet in **1** centered at δ 3.75 shifted downfield to δ 4.71 in **2**, confirming that the C-20 hydroxyl group is free and that the sugar in **1** is glycosidically linked to the C-3 hydroxyl group of the genin.

In the light of the foregoing evidence the structure of calocinin [**1**] is established as 3-O- β -L,2,6 dideoxy fucopyranoside.

EXPERIMENTAL

The general procedures were the same as reported earlier (8), except that the optical rotations were measured on a Perkin Elmer 241 automatic polarimeter, the 400-MHz ^1H nmr spectra on a Bruker instrument, and the mass spectra on an AEI MS-30 mass spectrometer. The plant *P. calophylla* was collected on Dec. 12, 1985 from Kempti Falls, Dehradun, Uttar Pradesh, India and identified by Dr. S.L. Kapoor, National Botanical Research Institute, Lucknow. A voucher specimen, Herbarium No. 116828, is preserved at the Institute.

PLANT EXTRACTION.—The extraction of the shade-dried, powdered twigs of *P. calophylla* was carried out as described earlier (8) and shaken with petroleum ether, Et_2O , CHCl_3 , CHCl_3 - EtOH (4:1), and CHCl_3 - EtOH (3:2) to fractionate its glycosides of different polarities. Repeated cc of CHCl_3 - EtOH (4:1 and 3:2) extracts over Si gel using $\text{CHCl}_3/\text{MeOH}$ as eluent afforded calocinin [**1**] (55 mg).

CALOCININ [1].—Mp 250–255° (MeOH/ Et_2O), $[\alpha]_{\text{D}}^{+25}$ +16° ($c = 0.13$, MeOH); found C 69.79, H 9.41; $\text{C}_{27}\text{H}_{44}\text{O}_6$ required C 69.83, H

9.48%. It gave a pink color in xanthydrol and blue color in the Keller-Kiliani reaction and also underwent NaIO_4 oxidation. $\text{IR } \nu_{\text{max}}$ (KBr) cm^{-1} 3490–3400, 2940, 1440, 1390, 1130, 1070, 820, 600; ^1H nmr (pyridine- d_5) δ 5.32 (1H, m, H-6), 4.64 (1H, dd, $J = 10$ and 2.5 Hz, H-1'), 3.75 (1H, q, H-20), 3.65 (1H, m, H-3'), 3.32 (1H, m, H-5'), 3.15 (1H, t, H-4'), 2.25–2.14 (1H, m, H-2'eq), 1.72–1.68 (1H, m, H-2'ax), 1.64–1.50 (methylene of aglycone), 1.34 (3H, d, $J = 6$ Hz, 6'-Me), 1.30 (3H, d, $J = 6$ Hz, 21-Me), 1.24 (3H, s, 18-Me), 1.02 (3H, s, 19-Me); ms m/z (rel. int.) $[\text{M}]^+$ (not observed) $[\text{M} - \text{CHOHMe}]^+ 419$ (11), $[\text{M} - \text{sugar}]^+ 334$ (15), $[\text{M} - \text{CHOHMe} - \text{sugar}]^+ 289$ (35), $[\text{289} - \text{H}_2\text{O}]^+ 271$ (30), $[\text{271} - \text{H}_2\text{O}]^+ 253$ (25); genin fragments $[\text{249} - \text{2H}_2\text{O}]^+ 213$ (9), $[\text{137} - \text{H}_2\text{O}]^+ 119$ (8); sugar fragments 148 (2), $[\text{148} - \text{OH}]^+ 131$ (100), $[\text{131} - \text{H}_2\text{O}]^+ 113$ (55), $[\text{113} - \text{H}_2\text{O}]^+ 95$ (15), $[\text{148} - \text{C}_3\text{H}_6\text{O} - \text{H}_2\text{O}]^+ 72$ (20), $[\text{148} - \text{H}_2\text{O} - \text{MeCHO} - \text{H}_2\text{O}]^+ 68$ (25).

TRI-O-ACETYLCALOCININ [2].—Compound **1** (12 mg) on acetylation with Ac_2O (0.15 ml) in pyridine (1.1 ml) at room temperature and usual workup yielded **2** as an amorphous residue (12 mg), $[\alpha]^{25}_{\text{D}} + 12^\circ$ ($c = 0.08$, CHCl_3) which failed to crystallize. ^1H nmr δ 5.38 (1H, m, H-6), 4.71 (1H, q, H-20), 4.64 (1H, dd, $J = 10$ and 2.5 Hz, H-1'), 4.21 (1H, t, H-4'), 2.05 (3H, s, OAc), 2.03 (3H, s, OAc), 2.02 (3H, s, OAc), 1.32 (3H, d, $J = 6$ Hz, 6'-Me), 1.22 (3H, d, $J = 6$ Hz, 21-Me), 1.02 (3H, s, 18-Me), 0.9 (3H, s, 19-Me).

MILD ACID HYDROLYSIS OF CALOCININ [1].—To a solution of **1** (25 mg) in 80% aqueous dioxane (1.7 ml) was added 0.025 M H_2SO_4 (1.7 ml), and the solution was warmed for 30 min at 50°. Dioxane was removed under reduced pressure, and the aqueous portion was repeatedly extracted with CHCl_3 -MeOH (99:1). The organic layer was washed in sequence with H_2O , 2 N Na_2CO_3 , again with H_2O , dried over Na_2SO_4 , and evaporated to afford genin **3** as colorless needles (11 mg), mp 198–201° (MeOH/Et₂O), $[\alpha]^{25}_{\text{D}} - 50^\circ$ ($c = 0.15$, MeOH); found C 75.38, H 10.10; $\text{C}_{21}\text{H}_{34}\text{O}_3$ required C 75.45, H 10.18%. It was identified as calogenin [lit. (8) mp 202–205°] by mmp and tlc comparison with the authentic sample. ^1H nmr δ 5.35 (1H, m, H-6), 3.84 (1H, q, $J = 6$ Hz, H-20), 3.51 (1H, m, H-3), 1.2 (3H, d, $J = 6$ Hz, 21-Me), 1.1 (3H, s, 18-Me), 0.75 (3H, s, 19-Me).

The aqueous hydrolyzate was neutralized with freshly prepared BaCO_3 , filtered, and concentrated under reduced pressure to afford the syrupy sugar **5** (7 mg): $[\alpha]^{25}_{\text{D}} - 58.5^\circ$ ($c = 1.0$, ω), which gave a pink coloration in xanthydrol and blue coloration in the Keller-Kiliani reaction and reacted with NaIO_4 . Sugar **5** was identified as 2-

deoxy-L-fucose ($[\alpha]_{\text{D}} - 60.2^\circ$) by comparing its relative mobility with that of digitoxose.

DI-O-ACETYL CALOGENIN [4].—Crystalline **3** (4 mg) acetylated in the usual manner afforded acetate **4** (4 mg): mp 163–165°; $[\alpha]^{25}_{\text{D}} - 18^\circ$ ($c = 0.3$, MeOH); found C 71.68, H 9.01; $\text{C}_{25}\text{H}_{38}\text{O}_5$ required C 71.77, H 9.09%.

OXIDATION OF 2-DEOXY-L-FUCOSE [5] WITH Br_2 WATER.—A solution of **5** (6 mg) in H_2O (0.4 ml) was mixed with Br_2 (6 μl), and the mixture was shaken in a stoppered flask in the dark for 24 h at room temperature. Excess Br_2 was then removed under reduced pressure, and the acidic mixture was made neutral with freshly precipitated Ag_2CO_3 . H_2S was passed through the filtrate to remove Ag^+ ions. This filtrate was evaporated to dryness under reduced pressure yielding L-fuconic acid lactone (4.5 mg), $[\alpha]^{25}_{\text{D}} + 28.5^\circ$ ($c = 1.0$, Me_2CO), showing a violet coloration with $\text{NH}_2\text{OH}\cdot\text{FeCl}_3$ spray reagent (19).

L-FUCONIC ACID PHENYLHYDRAZIDE.—A solution of L-fuconic acid lactone (4.5 mg) in absolute EtOH (0.06 ml) was heated with phenylhydrazine (0.06 ml) at 100° for 30 min. The cooled viscous mass was triturated with absolute Et₂O (to remove excess phenylhydrazine), yielding L-fuconic acid phenylhydrazide (3 mg): mp 165–168° (MeOH/Me₂CO), $[\alpha]^{25}_{\text{D}} - 7.0^\circ$ ($c = 1.0$, ω); found N 10.96; $\text{C}_{12}\text{H}_{18}\text{N}_2\text{O}_4$ required N 11.02%.

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LITERATURE CITED

1. T. Reichstein, *Naturwissenschaften*, **54**, 53 (1967).
2. R. Tschesche, *Bull. Soc. Chim. Fr.*, 1219 (1965).
3. M. Takase, S. Tarada, H. Yamamoto, T. Narita, M. Kimura, K. Hayashi, and H. Mitsuhashi, *Chem. Pharm. Bull.* **30**, 2429 (1982).
4. S. Yoshimura, H. Narita, K. Hayashi, and H. Mitsuhashi, *Chem. Pharm. Bull.* **31**, 3971 (1983).
5. D. Deepak, M.P. Khare, and A. Khare, *Indian J. Chem.*, **25**, 44 (1986).
6. D. Deepak, M.P. Khare, and A. Khare, *Phytochemistry*, **24**, 1037 (1985).
7. D. Deepak, M.P. Khare, and A. Khare, *Phytochemistry*, **24**, 3015 (1985).
8. O.P. Srivastava, A. Khare, and M.P. Khare, *J. Nat. Prod.*, **45**, 211 (1982).

9. O.P. Srivastava, A. Khare, and M.P. Khare, *J. Nat. Prod.*, **46**, 458 (1983).
10. E. Abisch and T. Reichstein, *Helv. Chim. Acta.* **43**, 1844 (1960).
11. G.M. Barton, R.S. Evans, and J.A.F. Gardner, *Nature*, **170**, 249 (1952).
12. R. Tschesche, G. Grimmer, and F. Seehofer, *Chem. Ber.*, **86**, 1235 (1953).
13. W. Nagata, C. Tamm, and T. Reichstein, *Helv. Chim. Acta.* **40**, 41 (1957).
14. S. Rangaswami and T. Reichstein, *Helv. Chim. Acta.* **32**, 939 (1949).
15. F. Schaub, H. Kaufmann, W. Stocklin, and T. Reichstein, *Helv. Chim. Acta.* **51**, 738 (1968).
16. B. Iselin and T. Reichstein, *Helv. Chim. Acta.* **27**, 1200 (1944).
17. H. Allgeier, *Helv. Chim. Acta.* **51**, 668 (1968).
18. P. Brown, F. Bruschweiler, G.R. Pettit, and T. Reichstein, *Org. Mass Spectrom.*, **5**, 573 (1971).
19. M. Abdel-Akher and F. Smith, *J. Am. Chem. Soc.* **73**, 5859 (1951).

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